



The Effects of ZnO Nanoparticles and ZnO/Chitosan NCs on Liver Histology and Serum Parameters in Rats

Hajinezhad MR

Associate Professor of physiology, Basic Science Department, Faculty of Veterinary Medicine, University of Zabol, Zabol, Iran.

Received: 20/06/2019

Accepted: 02/08/2019

Published: 01/12/2019

Abstract

There are conflicting and confusing reports about prooxidant/ activities of Zinc oxide nanoparticles and Zinc oxide NCs. This study aimed to resolve these discrepancies by examining the effects of these compounds on liver histopathology in healthy rats. Materials and Methods: 42 adult male Wistar rats were divided into 7 groups. Rats in the treatment group received intraperitoneal injections of ZnO nanoparticles 10, 20 and 40 mmol/ml and ZnO nanocomposites (NCs) (10, 20 and 40 mmol/ml) for 28 days. Control rats received distilled water. At the end of the study, the following parameters were assessed: serum liver enzymes (ALT and AST), the activity of serum catalase (CAT) and superoxide dismutase (SOD), serum BUN and serum creatinine and liver histology. Intraperitoneal injection of ZnO nanoparticles at a concentration of 10 mmol/ml/day had no significant effect on serum liver enzymes but at 20 and 40 mmol/ml/day significantly decreased serum catalase and SOD activity compared with the control group ($P < 0.05$). ZnO NCs at the concentrations of 20 and 40 mmol/ml/day decreased serum catalase activity and SOD activities and significantly elevates serum liver enzymes. Furthermore, both ZnO nanoparticles and ZnO NCs had no significant effect on serum BUN and creatinine levels. Both nanoparticles induced severe histological changes at the two higher doses (20 and 40 mmol/ml). The results suggest that proper concentrations of ZnO nanoparticles and ZnO NCs have no toxic effects on the liver while the higher doses can induce severe histological changes.

Keywords: ZnO nanoparticles, ZnO NCs, serum biochemical parameters, Rat

1 Introduction

Zinc is an essential component of many cytoplasmic enzymes that participates in the metabolism of carbohydrates, cytoplasmic proteins, lipids, and nucleic acids. This element induces many biological, physiological, and behavioral effects (1). Zinc is a necessary element for the growth of the body and zinc deficiency is associated with impaired immune function, irritability and stunting of growth. Specific age groups including infants, children, and adults are more susceptible to Zinc deficiency (2). Zinc is stable in the environment but is seldom abundant in food chains. Natural sources of zinc include meats, fruits, legume, and seeds. Major anthropogenic sources of zinc in the world are usually arising from municipal, industrial and agricultural activities (3). The brain, hematopoietic, hepatic and renal systems are the most important organs affected by Zinc deficiency (4). Nowadays there is an increasing trend toward the use of Zn in nanotechnology in the form of Zinc nanoparticles. Nanotechnology is a field of science which is about the preparation of nanoscale materials (5-7). Nanoparticles exhibit new and improved characteristics such as morphology and size distribution respecting larger particles from which they are made (8-25). Zinc nanoparticles are widely used in agriculture, cosmetic industry and medicine. ZnO nanoparticles, due to their

unique optical, chemical, mechanical and electrical properties, have attracted scientific attention (26). Based on previous reports, ZnO NPs can increase or decrease reactive oxygen species (ROS) generation and subsequently oxidative stress in different parts of the body (27). Reactive oxygen species are short-lived organic compounds with uneven number of electrons that stabilize themselves by oxidizing biological molecules (28). Normally, the reactive oxygen-containing molecules and other free radicals can be quickly removed by natural defense mechanisms such as Glutathione peroxidase, superoxide dismutase, and catalase (29). Nanocomposites are usually synthesized by using ionic gelation of pentasodium tripolyphosphate and chitosan. Applications of these compounds have also become more widespread regarding their chemical properties (30). Adding chitosan to Zinc structure affects its biological and physicochemical properties (31) So in the present study, we decided to examine whether adding chitosan to zinc can decrease or increase the and pro-oxidant activities of Zinc.

2 Material and Methods

2.1 Animals

Wistar rats (201- 234 g) obtained from Laboratory Animal Center University of Zabol were used in the current study. Animals were maintained in well-ventilated

Corresponding author: Hajinezhad MR. Associate Professor of physiology, Basic Science Department, Faculty of Veterinary Medicine, University of Zabol, Zabol, Iran.

rooms at a constant temperature of 20–23 °C and 12 (h) light/dark cycles with free to standard rodent food (Javaneh-Khorasan, Iran) and tap water. they were handled in accordance with the Animal Ethics Committee of the University of Zabol, Zabol- Iran, and Guide for the Care and Use of Laboratory Animals (National Institutes of Health (NIH) publication 86–23; revised 1985. 42 adult male Wistar rats were divided into 8 groups (7 rats in each group: one control group and six treatment group. Rats in treatment group received intraperitoneal injections of ZnO nanoparticles (10, 20 and 40 mmol/ml/day) and ZnO NCs (10, 20 and 40 mmol/ml/day) for 28 days. Control rats received 1 ml of distilled water intraperitoneally for 28 days. At the end of our study, blood samples were collected via retro-orbital puncture and centrifuged (3000 rpm for 10 minutes) for serum separation. The serum samples were immediately frozen at –80° C. Following blood collection, rats were sacrificed by cervical dislocation and whole-brain tissues were isolated. The fresh brain tissues were immediately washed with 0.9% Na Cl and stored at –20°C for further determination of lipid peroxidation in MDA form.

3.1 Serum biochemical parameters

Analyses of serum ALT, AST and ALP levels were performed by using the Selectra pro, M auto analyzer, (Vital Scientific, SpanNeren, Netherlands) with Pars Azmoon reagents kit (Pars Azmoon. Co., Tehran, Iran). Serum creatinine and BUN were measured using commercial kits (Pars Azmoon Lab, Iran), according to manufacturer's instructions. Serum antioxidant enzymes were measured using commercial kits (Zell Bio Germany) and according to the company instructions.

3.2 Histopathological examination

After animals were euthanized by diethyl ether, liver specimens were separated and washed in water. The liver specimens were then sliced and preserved in a 10% neutral buffered formalin solution (NBF). After paraffin

embedding and block making of tissue samples, serial sections were prepared by using the hematoxylin-eosin method and were examined under a light microscope (Olympus, Tokyo, Japan).

2.2 Statistical analysis

The collected data were analyzed with SPSS software (version 20.0) expressed as mean $\pm$ SD. All Multiple comparisons were performed by using one-way analysis of variance (ANOVA) followed by posthoc Tukey's test. Statistical significance was set at $P < 0.05$.

3 Results

Intraperitoneal injection of ZnO NPs at a concentration of 10 mmol/ml had no significant effect on serum biochemical parameters ($P > 0.05$). Also, liver histology was not affected by this dose (fig.2). However, ZnO NPs at 20 and 40 mmol/ml significantly decreased serum catalase and superoxide dismutase activity compared with the control group ($P < 0.05$) (fig.1). As seen in fig.2, serum liver enzymes ALT and AST were increased by ZnO NPs administration ($P > 0.05$). Furthermore, intraperitoneal injection of two low concentrations of ZnO NPs did not affect serum creatinine and BUN fig.3. As seen in fig 3, ZnO NPs intraperitoneal injection at a concentration of 40 mmol/ml caused a non-significant difference in kidney function tests (BUN and serum creatinine) fig.3 ($P > 0.05$). ZnO NCs at a concentration of 40 mmol/ml decreased serum catalase and SOD activity ($P < 0.05$). Intraperitoneal injection of ZnO NCs also increased serum liver enzymes (fig 2), while serum BUN and creatinine levels were not affected by ZnO NCs administration fig.3. As expected, ZnO nanoparticles and ZnO NCs had no significant effect on serum BUN and creatinine fig.5. Liver histological investigation of rats received a low dose of ZnO NCs showed normal architecture. However, the rats received the 20 mmol/ml and 40 mmol/ml of NCs showed signs of necrosis and hemorrhage (fig.5).

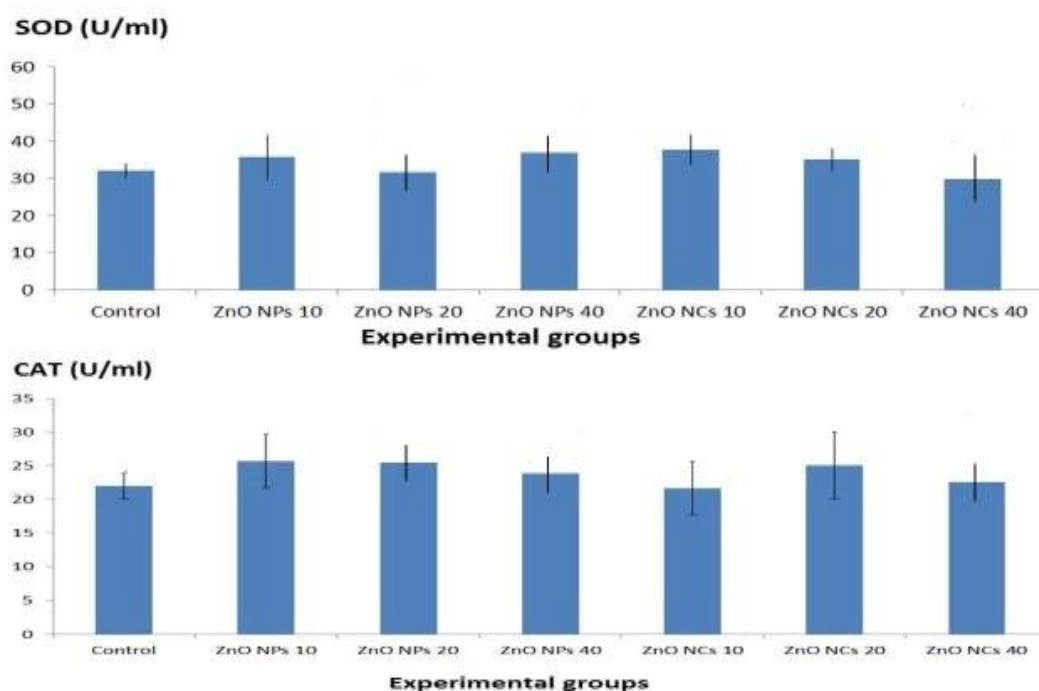


Fig. 1: Serum CAT and SOD activities in rats treated with ZnO NPs ($\bar{X} \pm SD$, $n=10$). * $P < 0.05$, compared to control group.

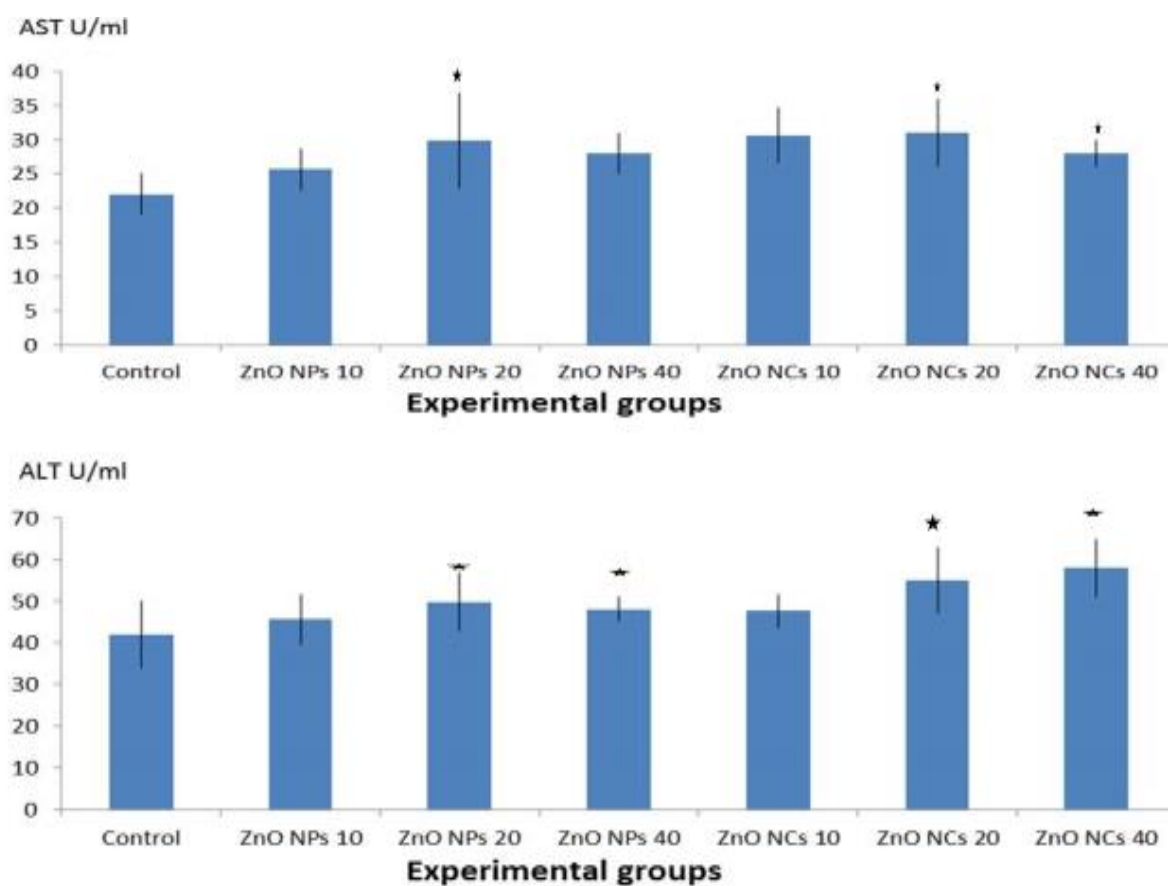


Fig. 2: Serum ALT and AST activities in rats treated with *ZnO NPs* ($X \pm SD$, $n=10$). * $P < 0.05$ compared to control group

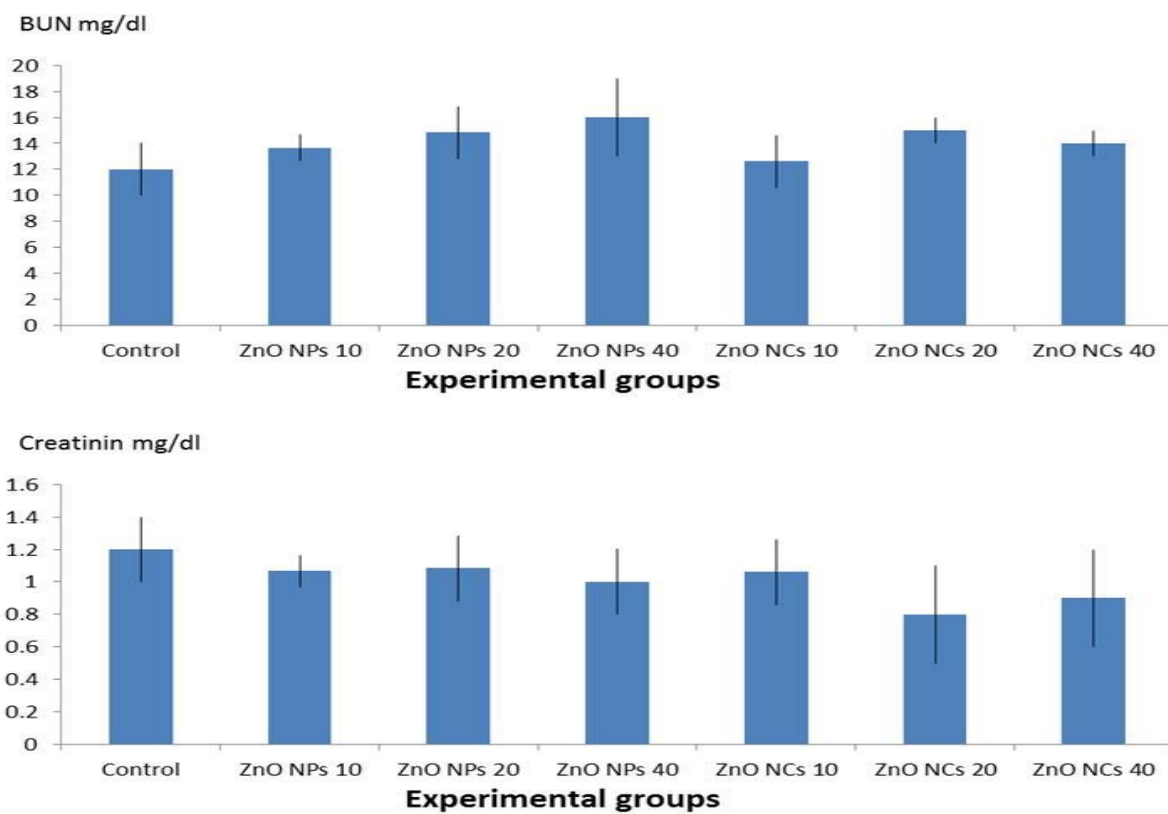


Fig. 3: Serum BUN and creatinine levels in rats treated with *ZnO NPs* ($X \pm SD$, $n=10$). * $P < 0.05$ compared to control group.

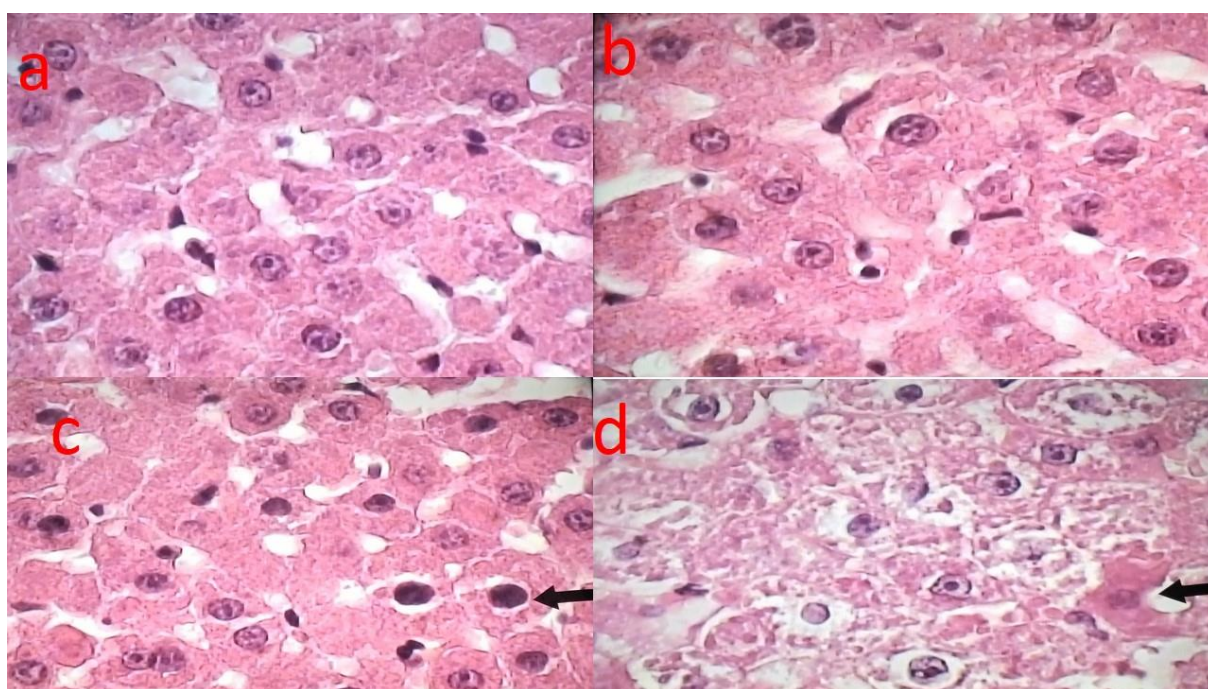


Fig. 4: Liver histology in control rats (a): Normal liver architecture with intact hepatocytes and sinusoids, rats received ZnO nanoparticles 10 mmol (b): normal hepatocytes, rats received ZnO nanoparticles 20 mmol (c): necrosis (arrow), rats received ZnO nanoparticles 30 mmol (d): necrosis (arrow) and fatty changes. . H&E staining ($\times 40$)

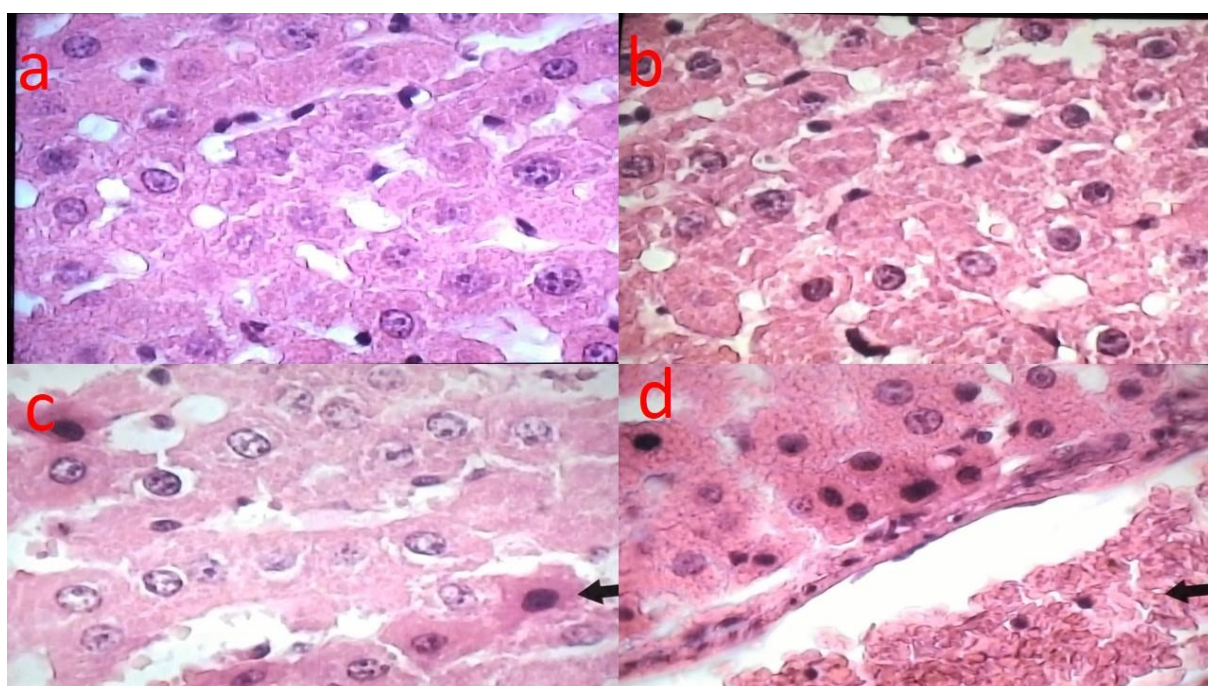


Fig. 5: Liver histology in control rats (a): healthy hepatocytes and well-arranged sinusoids, rats received ZnO nanoparticles 10 mmol (b): normal liver histological pattern, rats received ZnO nanoparticles 20 mmol (c): necrosis (arrow), rats received ZnO nanoparticles 30 mmol (d): necrosis (arrow) and bleeding. . H&E staining ($\times 40$).

4 Discussion

The effects of ZnO NPs on serum oxidative stress biomarkers have not been fully elaborated yet. Some studies had reported the properties of ZnO nanoparticles (32) while other studies indicated oxidative damage caused by ZnO nanoparticles (33). As seen in this study, administration of ZnO nanoparticles decreased serum enzyme activities. These results were in line with previous studies about the biological effects of ZnO nanoparticles (34, 35) It can be concluded that the decreases in serum

catalase and superoxide dismutase activity is an indicator of pro-oxidant activities of ZnO nanoparticles (36). Superoxide radical anion, peroxyl radicals which are the major reactive oxygen species generated during oxidative stress, may induce liver fibrosis and fatty changes by activating type I procollagen synthesis enzyme. This process may lead to the generation of lipid peroxidation end-products. Our eukaryotic cells have a broad types of oxidative defense mechanisms, like enzymatic molecules such as CAT, SOD, glutathione peroxidase and molecular scavengers such as vitamin A, reduced

glutathione (GSH), ascorbic acid, vitamin E (alpha tocopherol) and N-acetyl-5-methoxytryptamine (melatonin) (37). In this investigation, decrease in the activities of SOD and CAT might be due to increased generation of reactive oxygen species or increased availability of NADPH that is required to maintain oxidative defenses mechanisms. These results suggest that the ZnO nanoparticles have free radical producing effect and can reduce the activity of the endogenous enzymes. In this study, a significant increase in serum AST and ALT levels was observed in rats treated with ZnO nanoparticles. These results were in line with other studies indicating hepatic damage caused by ZnO nanoparticles or with studies indicating hepatoprotective potential of these nanoparticles (38, 39). As previously noted, damage to liver tissues releases the AST, ALP and ALT into the serum, and, hence, elevation of serum activities of these enzymes is considered a valuable marker of liver damage (40). Serum creatinine and blood urea nitrogen are used as indicators of glomerular filtration rate and renal function (41). ZnO nanoparticles or ZnO nanochitosans had no significant effect on serum kidney function markers. These findings were in contrast with previous studies that reported significant increase in serum kidney markers following treatment with ZnO nanoparticles (8-23, 42). This result might be due to the low doses of ZnO nanoparticles or ZnO nanochitosans used in this study. This observation led us to conclude that low concentrations of these compounds have potential without any toxicity in vivo.

5 Conclusion

Our results indicate that ZnO nanoparticles have similar effects compared to ZnO NCs. Further studies are required to elucidate the effects of these compounds in laboratory animals.

Acknowledgment

This study was based on the grant on University of Zabol Grant number: 9618-15. We are grateful to University of Zabol for financial support.

Conflicts of interest

There are no conflicts of interest.

References

- Kambe T, Tsuji T, Hashimoto A, Itsumura N. The physiological, biochemical, and molecular roles of zinc transporters in zinc homeostasis and metabolism. *Physiological reviews*. 2015;95(3):749-84.
- Prasad AS. Discovery of human zinc deficiency: its impact on human health and disease. *Advances in nutrition*. 2013;4(2):176-90.
- Jung SK, Kim M-K, Lee Y-H, Shin DH, Shin M-H, Chun B-Y, et al. Lower zinc bioavailability may be related to higher risk of subclinical atherosclerosis in Korean adults. *PLoS One*. 2013;8(11):e80115.
- Doboszewska U, Szewczyk B, Sowa-Kućma M, Noworyta-Sokołowska K, Misztak P, Gołębiowska J, et al. Alterations of bio-elements, oxidative, and inflammatory status in the zinc deficiency model in rats. *Neurotoxicity research*. 2016;29(1):143-54.
- Salari Z, Ameri A, Forootanfar H, Adeli-Sardou M, Jafari M, Mehrbani M, et al. Microwave-assisted biosynthesis of zinc nanoparticles and their cytotoxic and antioxidant activity. *Journal of Trace Elements in Medicine and Biology*. 2017;39:116-23.
- Alaee S, Ilani M. Effect of titanium dioxide nanoparticles on male and female reproductive systems. *Journal of Advanced Medical Sciences and Applied Technologies*. 2017;3(1):3-8.
- Ravanshad R, Karimi Zadeh A, Amani AM, Mousavi SM, Hashemi SA, Savar Dashtaki A, et al. Application of nanoparticles in cancer detection by Raman scattering based techniques. *Nano reviews & experiments*. 2018;9(1):1373551.
- Amani A. Synthesis and biological activity of piperazine derivatives of phenothiazine. *Drug Res*. 2015;65(01):5-8.
- Beheshtkhoo N, Kouhbanani MAJ, Savardashtaki A, Amani AM, Taghizadeh S. Green synthesis of iron oxide nanoparticles by aqueous leaf extract of *Daphne mezereum* as a novel dye removing material. *Appl Phys A*. 2018;124(5):363.
- Hashemi SA, Mousavi SM, Faghihi R, Arjmand M, Sina S, Amani AM. Lead oxide-decorated graphene oxide/epoxy composite towards X-Ray radiation shielding. *Radiation Physics and Chemistry*. 2018;146:77-85.
- Kouhbanani MAJ, Beheshtkhoo N, Amani AM, Taghizadeh S, Beigi V, Bazmandeh AZ, et al. Green synthesis of iron oxide nanoparticles using *Artemisia vulgaris* leaf extract and their application as a heterogeneous Fenton-like catalyst for the degradation of methyl orange. *Materials Research Express*. 2018;5(11):115013.
- Kouhbanani MAJ, Beheshtkhoo N, Fotoohiardakani G, Hosseini-Nave H, Taghizadeh S, Amani AM. Green Synthesis and Characterization of Spherical Structure Silver Nanoparticles Using Wheatgrass Extract. *Journal of Environmental Treatment Techniques*. 2019;7(1):142-9.
- Kouhbanani MAJ, Beheshtkhoo N, Taghizadeh S, Amani AM, Alimardani V. One-step green synthesis and characterization of iron oxide nanoparticles using aqueous leaf extract of *Teucrium polium* and their catalytic application in dye degradation. *Advances in Natural Sciences: Nanoscience and Nanotechnology*. 2019;10(1):015007.
- Lohrasbi S, Kouhbanani MAJ, Beheshtkhoo N, Ghasemi Y, Amani AM, Taghizadeh S. Green Synthesis of Iron Nanoparticles Using *Plantago major* Leaf Extract and Their Application as a Catalyst for the Decolorization of Azo Dye. *BioNanoScience*. 2019;9(2):317-22.
- Mahdavinia GH, Rostamizadeh S, Amani AM, Sepehrian H. Fast and efficient method for the synthesis of 2-arylbenzimidazoles using MCM-41-SO₃H. *Heterocycl Commun*. 2012;18(1):33-7.
- Mousavi SM, Hashemi SA, Arjmand M, Amani AM, Sharif F, Jahandideh S. Octadecyl amine functionalized Graphene oxide towards hydrophobic chemical resistant epoxy Nanocomposites. *ChemistrySelect*. 2018;3(25):7200-7.
- Rostamizadeh S, Amani AM, Aryan R, Ghaieni HR, Norouzi L. Very fast and efficient synthesis of some novel substituted 2-arylbenzimidazoles in water using ZnOCl₂·nH₂O on montmorillonite K10 as catalyst. *Monatshefte für Chemie-Chemical Monthly*. 2009;140(5):547-52.
- Rostamizadeh S, Amani AM, Mahdavinia GH, Shadjou N. Silica supported ammonium dihydrogen phosphate (NH₄H₂PO₄/SiO₂): A mild, reusable and highly efficient heterogeneous catalyst for the synthesis of 14-aryl-14-H-dibenzo [a, j] xanthenes. *Chinese Chemical Letters*. 2009;20(7):779-83.
- Rostamizadeh S, Aryan R, Ghaieni HR, Amani AM. Aqueous NaHSO₄ catalyzed regioselective and versatile synthesis of 2-thiazolamines. *Monatshefte für Chemie/Chemical Monthly*. 2008;139(10):1241-5.
- Rostamizadeh S, Aryan R, Ghaieni HR, Amani AM. An efficient one-pot procedure for the preparation of 1, 3, 4-thiadiazoles in ionic liquid [bmim] BF₄ as dual solvent and catalyst. *Heteroatom Chemistry: An International Journal of Main Group Elements*. 2008;19(3):320-4.
- Rostamizadeh S, Aryan R, Ghaieni HR, Amani AM. Solvent-free chemoselective synthesis of some novel substituted 2-arylbenzimidazoles using amino acid-based prolinium nitrate ionic liquid as catalyst. *Journal of Heterocyclic Chemistry*. 2009;46(1):74-8.

22. Rostamizadeh S, Aryan R, Reza Ghaieni H, Mohammad Amani A. Efficient synthesis of 1, 3, 4-thiadiazoles using hydrogen bond donor (thio) urea derivatives as organocatalysts. *J Heterocycl Chem*. 2010;47(3):616-23.
23. Zakeri A, Kouhbanani MAJ, Beheshtkhoo N, Beigi V, Mousavi SM, Hashemi SAR, et al. Polyethylenimine-based nanocarriers in co-delivery of drug and gene: a developing horizon. *Nano reviews & experiments*. 2018;9(1):1488497.
24. Mohammad Amin Jadidi Kouhbanani † a NBA, Pourya Nasirmoghadas b , Samira Yazdanpanah c , Kamiar Zomorodianc, Saeed Taghizadeh d, Ali Mohammad Amani. Green Synthesis of Spherical Silver Nanoparticles Using *Ducrosia Anethifolia* Aqueous Extract and Its Antibacterial Activity. *Journal of Environmental Treatment Techniques (JETT)*. 2019;;7(3):461-6.
25. Mousavi SM, Hashemi SA, Amani AM, Saed H, Jahandideh S, Mojoudi F. Polyethylene terephthalate/acryl butadiene styrene copolymer incorporated with oak shell, potassium sorbate and egg shell nanoparticles for food packaging applications: control of bacteria growth, physical and mechanical properties. *Polymers from Renewable Resources*. 2017;8(4):177-96.
26. Kim SW, Nguyen TK, Van Thuan D, Dang DK, Hur SH, Kim EJ, et al. Polyol-mediated synthesis of ZnO nanoparticle-assembled hollow spheres/nanorods and their photoanode performances. *Korean Journal of Chemical Engineering*. 2017;34(2):495-9.
27. Huang C-C, Aronstam RS, Chen D-R, Huang Y-W. Oxidative stress, calcium homeostasis, and altered gene expression in human lung epithelial cells exposed to ZnO nanoparticles. *Toxicology in vitro*. 2010;24(1):45-55.
28. Amani A. Synthesis, characterization and antibacterial and antifungal evaluation of some para-quinone derivatives. *Drug Res*. 2014;64(08):420-3.
29. Pisoschi AM, Pop A. The role of antioxidants in the chemistry of oxidative stress: A review. *European journal of medicinal chemistry*. 2015;97:55-74.
30. Abdel-Wahhab MA, Aljawish A, El-Nekeety AA, Abdel-Aziem SH, Hassan NS. Chitosan nanoparticles plus quercetin suppress the oxidative stress, modulate DNA fragmentation and gene expression in the kidney of rats fed ochratoxin A-contaminated diet. *Food and Chemical Toxicology*. 2017;99:209-21.
31. Deshpande P, Dapkekar A, Oak MD, Paknikar KM, Rajwade JM. Zinc complexed chitosan/TPP nanoparticles: a promising micronutrient nanocarrier suited for foliar application. *Carbohydrate polymers*. 2017;165:394-401.
32. Muthukumar H, Pichiah S, Leong KH, Devi SA, Manickam M. Facile biosynthesis of ZnO and iron doped ZnO nanocatalyst: physicochemical traits and multifunctional applications. *Journal of Bionanoscience*. 2017;11(2):114-22.
33. Ng CT, Yong LQ, Hande MP, Ong CN, Yu LE, Bay BH, et al. Zinc oxide nanoparticles exhibit cytotoxicity and genotoxicity through oxidative stress responses in human lung fibroblasts and *Drosophila melanogaster*. *International journal of nanomedicine*. 2017;12:1621.
34. Das D, Nath BC, Phukon P, Dolui SK. Synthesis of ZnO nanoparticles and evaluation of antioxidant and cytotoxic activity. *Colloids and Surfaces B: Biointerfaces*. 2013;111:556-60.
35. Nagajyothi P, Sreekanth T, Tetley CO, Jun YI, Mook SH. Characterization, antibacterial, antioxidant, and cytotoxic activities of ZnO nanoparticles using *Coptidis Rhizoma*. *Bioorganic & medicinal chemistry letters*. 2014;24(17):4298-303.
36. Shoaie-Hagh P, Rahimifard M, Navaei-Nigjeh M, Baeeri M, Gholami M, Mohammadirad A, et al. Zinc oxide nanoparticles reduce apoptosis and oxidative stress values in isolated rat pancreatic islets. *Biological trace element research*. 2014;162(1-3):262-9.
37. Dalle-Donne I, Rossi R, Colombo R, Giustarini D, Milzani A. Biomarkers of oxidative damage in human disease. *Clinical chemistry*. 2006;52(4):601-23.
38. Pandurangan M, Kim DH. ZnO nanoparticles augment ALT, AST, ALP and LDH expressions in C2C12 cells. *Saudi journal of biological sciences*. 2015;22(6):679-84.
39. Yan G, Huang Y, Bu Q, Lv L, Deng P, Zhou J, et al. Zinc oxide nanoparticles cause nephrotoxicity and kidney metabolism alterations in rats. *Journal of Environmental Science and Health, Part A*. 2012;47(4):577-88.
40. Weemhoff JL, Woolbright BL, Jenkins RE, McGill MR, Sharpe MR, Olson JC, et al. Plasma biomarkers to study mechanisms of liver injury in patients with hypoxic hepatitis. *Liver International*. 2017;37(3):377-84.
41. Coca S, Yalavarthy R, Concato J, Parikh C. Biomarkers for the diagnosis and risk stratification of acute kidney injury: a systematic review. *Kidney international*. 2008;73(9):1008-16.
42. Wang D, Li H, Liu Z, Zhou J, Zhang T. Acute toxicological effects of zinc oxide nanoparticles in mice after intratracheal instillation. *International journal of occupational and environmental health*. 2017;23(1):11-9.